THE OXIRANE RING OPENINGS OF THE DIANHYDRO SUGAR WITH HIGH REGIOSELECTIVITY AND ITS USE IN PREPARATION OF TWO CHIRAL SEGMENTS OF 6-DEOXYERYTHRONOLIDE B

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Summary: The oxirane ring of the dianhydro sugar 2 obtained from levoglucosan (1,6-anhydro- β -D-glucose) is opened regioselectively with organometallic carbon nucleophiles and its application to an asymmetric synthesis of the C_1 - C_5 segment 9 together with the C_9 - C_{15} unit 18 of 6-deoxyerythronolide B 1 is described.

Many monosaccharides and a large number of their derivatives are easily available and versatile compounds. They prove to be now an important source of chirality for the total synthesis of optically active natural products.¹ The structurally rigid and readily available levoglucosan 4 has been used recently in the synthesis of biologically active natural products.² Most recent report³ that the 2,3-epoxide 2 was effectively opened trans-diaxially only with the reagent of the lithium anion of thioanisole as carbon nucleophiles prompted us to describe our own results, which provided a convenient method for the preparation of dialkyl alcohol 3 from 2 by reaction with organometallic reagents.



In connection with ongoing program directed toward utilization of levoglucosan 4 for synthesis of naturally occurring macrolide antibiotics, its application to the synthesis of two chiral segments of 6-deoxyerythronolide B 1 is described.



Levoglucosan **1**⁴ was converted, via the ditosylate, into the 3,4-epoxy-Compounds 2a and 2b were prepared in good yield by copper(1) tosvlate 5.⁵ induced reaction ^{2a,b} with methyl- or allylmagnesium chloride in THF, followed by treatment with sodium methoxide. A similar sequence for the preparation of 2c and 2d was applied to give rise to 62% yields respectively. Conversion of the 2, 3-epoxide 2 into the dialkylated anhydro sugar 3 was achieved with a preferential trans-diaxial opening of oxirane ring and a wide variety of organometallic reagents were attempted before suitable conditions for the transformation were found. The reactions as shown in Table 1 were carried out either by using copper(1) catalyzed additions of Grignard reagents or by employing dimethylcopperlithium at 0° C in THF to afford 3 in acceptable vield.

entry	R ¹	R ²	organometallic reagent	yield(%)
а	Me	Me	Me ₂ CuLi	41
b	Me	CH2=CHCH2	CH ₂ =CHCH ₂ MgBr /CuI	7 (^{b)}
с	$CH_2 = CHCH_2$	Me	MeMg I / Cu I	70
d	CH ₂ =CHCH ₂	CH2=CHCH2	CH ₂ =CHCH ₂ MgC1 /Cu1	56
е	Et	CH2=CHCH2	CH ₂ =CHCH ₂ MgBr /Cul	5 5
f	CH ₂ =CVieCH ₂	Me	MeMgBr/CuI	45

Table 1 Conversion of epoxide 2 to alcohol 3^{a}

The reactions were performed at 0°C in THF unless indicated.
 The reaction was carried out at room temperature and the regioisomer (trans diequatrial opening) was isolated in 13% yield.

In the remaining paragraph, we will describe the synthesis of the optically active segments $C_1^{-}C_5^{-}$ and $C_9^{-}C_{15}^{-}$ of 6-deoxyerythronolide B 1, which is a macrolide antibiotic with ten asymmetric centers.⁶

Compounds 3a and $6a(R^1=Me)$ can be used in the preparation of two units 9 and 18 of 1 as follows. Benzylation of 3a (PhCH₂Br/NaH/cat.n-Bu₄NI, 94%) followed by anhydro ring opening (CSA/MeOH) gave smoothly acetal 7, which in turn was subjected to thioacetalization (HS(CH₂)₃SH/ BF₃·Et₂O) to furnish thioacetal 8 in 65% yield. Oxidative cleavage of 8 with lead tetraacetate and reduction with sodium borohydride led to the corresponding alcohol 9 in 63% yield. Thioacetal 9 obtained in this way was also converted to lactone 10 via the following functional group manipulation involving (I) AgND₃/NCS (II) Jones oxidation (III) H₂/Pd(OH)₂-C. Lactone 10, which was obtained by Gerson⁷ from part of (9S)-dihydroerythronolide A, had mp 88°C, and [α]_D=-4.6° (MeOH)[lit.⁷ mp 88-89°C, [α]_D=-5.0°(MeOH)].



a) BzlBr/NaH/n-Bu₄NI(cat) b) MeOH/CSA c) HS(CH₂)₃SH/BF₃·Et₂O d) Pb(OAc)₄/C₆H₆ e) NaBH₄/MeOH f) AgNO₃/NCS g) Jones axid. h) H₂/Pd(OH)₂-C/MeOH

Turning next to elaboration of the segment 18 , the anhydro benzyl ether 11was easily prepared from $6a(R^1 = Me)$ in two steps ((1) LiBEt₂H/THF, $80\%^{2c}$ (11) C₆H₅CH₉Br/NaH/cat.n-Bu₄NI, 94%). Subsequent ring opening of 11 (MeOH/BF₂• $Et_{2}O$) gave an anomeric mixture of acetal 12 in quantitative yield. Transformation of 12 into 13 was performed by a conventional way in four steps ((I) $\label{eq:processed} \text{PCC/MS4A}^8,80\% \quad (\text{II}) \ \text{MeMgBr/THF,83\%} \ (\text{III}) \ \text{PCC/MS4A} \ (\text{IV}) \ \text{Ph}_3\text{P=CH}_2/\text{THF,67\%} \ \text{in two}$ Hydrolysis of 13 with 10% H_2SO_4 in THF and subsequent reduction with steps). LiAlH₄ gave diol 14 in 76% yield. Concomitant protection of the primary Epoxidation⁹ of the hydroxyl group of 14 afforded 15 in 81% yield. allylic alcohol 15 (VO(acac)₂/TBHP/C₆H₆) produced diastereomerically pure 16 in 79% yield. When epoxide $\tilde{16}$ was allowed to react with NaBH $_3$ CN 10 in the presence of $\mathrm{BF}_{3}\cdot\mathrm{Et}_{2}\mathrm{O}$, a 33% yield of diol 17 was obtained with the desired stereochemistry at C-10 position.¹¹



a) LIBEt₃H/THF b) BzIBr/NaH/n-Bu₄NI(cat) c) MeOH/BF₃·Et₂O d) PCC/MS4A e) MeMgBr/THF f) Ph₃P⁺CH₃Br⁻/n-BuLi/THF



g) 10%H₂SO₄/THF h) LiAlH₄/THF i)^tBuCOC1/Pyr. j) VO(acac)₂/TBHP/C₆H₆ k) NaBH₃CN/BF₃·Et₂O/THF i) Me₂C(OMe)₂/CSA m) LiAlH₄/Et₂O n) TsC1/Pyr. o) LiBEt₃H/THF References and Notes

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- Efforts to improve this yield are currently undertaken in our labolatory. Tetrahydrofuran derivative was isolated as by-product in this reaction.

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